

BYK-ET 3001

Wetting and dispersing additive for solvent-borne concentrates of electrically conductive carbons and for the formulation of highly filled electrode slurries of Li-ion cells.

Product data

Composition

Solution of polyether-modified styrene-maleic anhydride copolymer

Typical properties

The values indicated in this data sheet describe typical properties and do not constitute specification limits.

Density (20 °C):	1.07 g/ml
Non-volatile matter (20 min, 150 °C):	66 %
Solvent:	N-methylpyrrolidone
Acid value:	27 mg KOH/g
Amine value:	37 mg KOH/g

Storage and transportation

To be stored and transported between 10 °C and 50 °C. Separation or turbidity may occur at temperatures below 10 °C. Warm to 20 °C and stir.

Special note

Electrochemical stability: 0.1 V to 4.8 V (vs. Li/Li⁺)

Applications

Energy storage

Special features and benefits

The stabilizing effect of BYK-ET 3001 shortens the dispersion process of carbon black and other conductive carbons (e.g. Ketjenblack®, carbon nanotubes, etc.). Adding BYK-ET 3001 achieves better stabilization of the particles, leading to reduced viscosity of the slurry compared to standard systems without dispersant.

BYK-ET 3001 facilitates the formulation of electrode slurries for Li-ion cells with a higher solid content. The higher solid content in turn accelerates the drying process and improves the productivity of the electrode manufacturing.

Recommended levels

Amount of additive (as supplied) based upon:

Conductive carbons (carbon black/Ketjenblack®/CNT/graphene): 5–50 %

The above recommended levels can be used for orientation. The optimum dosage should be determined by application-related test series.

Incorporation and processing instructions

BYK-ET 3001 should first be mixed with the solvent. Then the carbons should be added and homogeneously mixed in. The suitable dispersion time of the slurry is determined by measuring the particle size development over the dispersion time. The volume resistivity of the electrode coating should also be measured over the dispersion time.



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